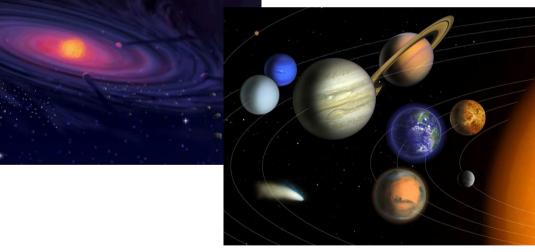
# ASTROCHEMISTRY & ASTROBIOLOGY: ISM, COMETS & METEORITES



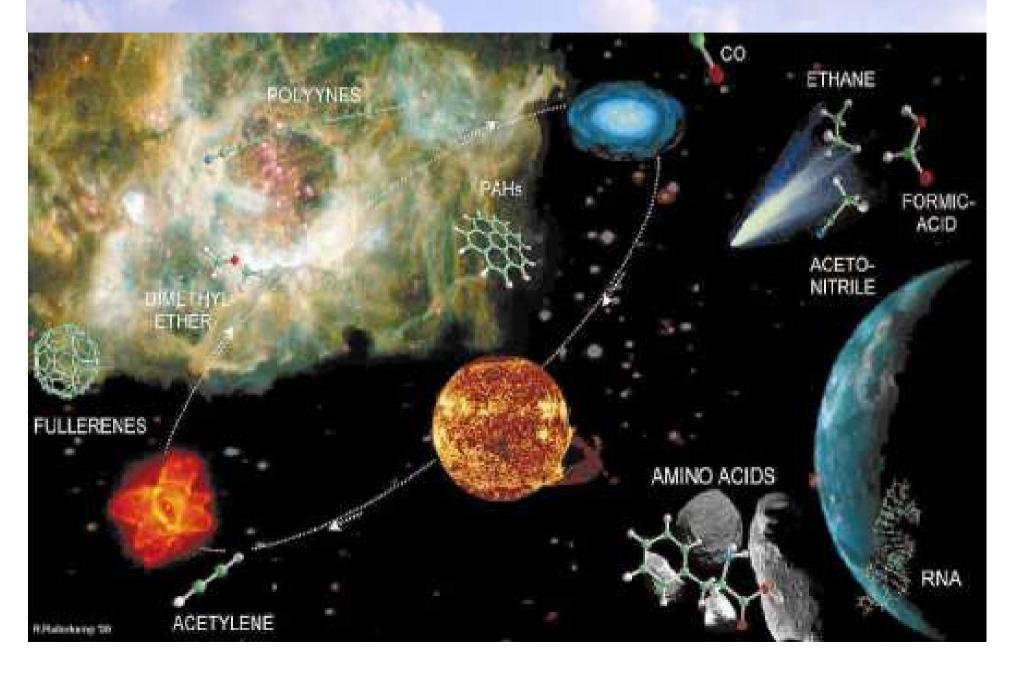
### Steven Charnley

NASA Ames Research Center



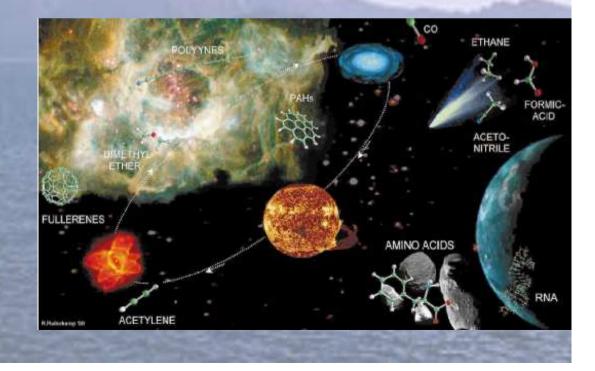
XII IAG/USP Advanced School on Astrophysics

# **Organic Molecules in the Universe**



# **Course Outline**

- 1. Overview Molecules in the ISM
- 2. Astrochemical Processes
- 3. Chemistry of Star Formation
- 4. Comets & Meteorites
- 5. Astrochemistry & Astrobiology



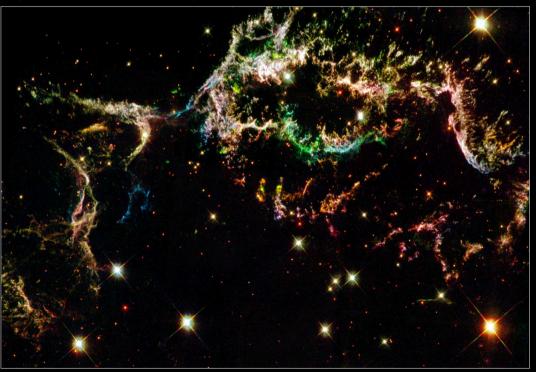
### **Cosmochemistry - Origin of the Elements and Dust**

Formation of biogenic elements in stellar nucleosynthesis. Injected into ISM by supernovae and stellar winds.

Injection of dust grains in SN ejecta and from atmospheres of late-type stars.



Supernova Remnant Cassiopeia A





NASA and The Hubble Heritage Team (STScl/AURA) • Hubble Space Telescope WFPC2 • STScl-PRC02-15

### **DENSE INTERSTELLAR CLOUDS**

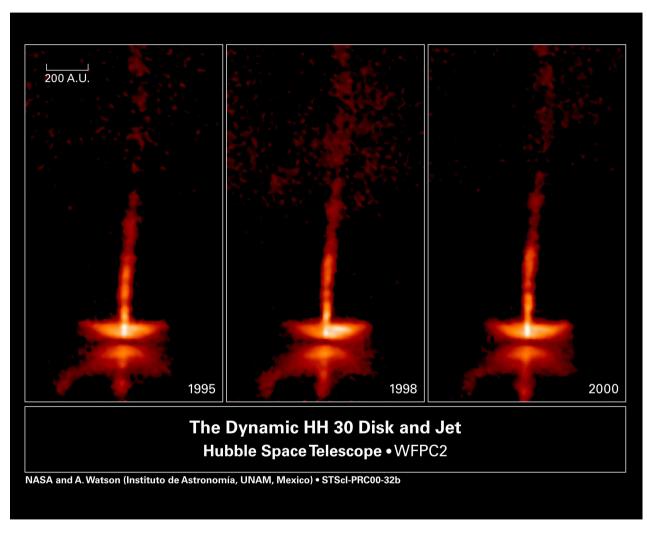
**NGC 604** 

- Almost all molecular H<sub>2</sub>
- Lifetime: 10<sup>6</sup> or 10<sup>7</sup> 10<sup>8</sup> years ?
- Site of star formation

Cosmic rays drive a rich ion-molecule chemistry supplemented by neutral-neutral processes organic molecules

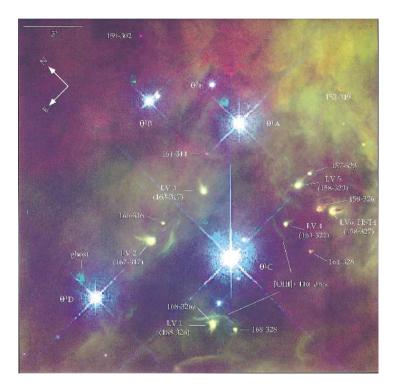
T ~ 10 K n~ 10<sup>3</sup> - 10<sup>6</sup> H atoms per cm<sup>3</sup>

# Flared Proto-planetary Disk in HH-30

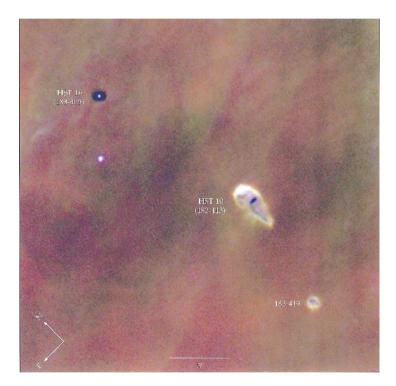


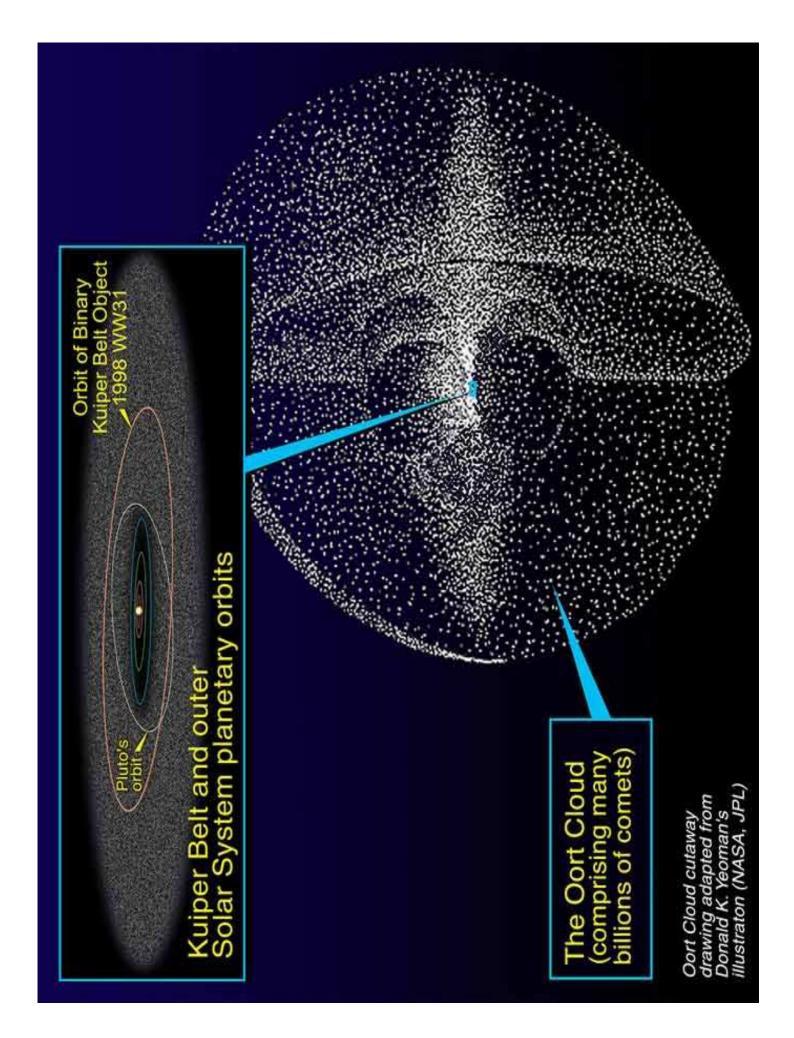
HH-30 is about 140 parsecs from Earth; 200 AU = 1.4 arc-sec on the sky

### Proplyds in Orion



After Bally et al. AJ 116:293 (1998)







## Asteroids



### **Carbonaceous meteorites**

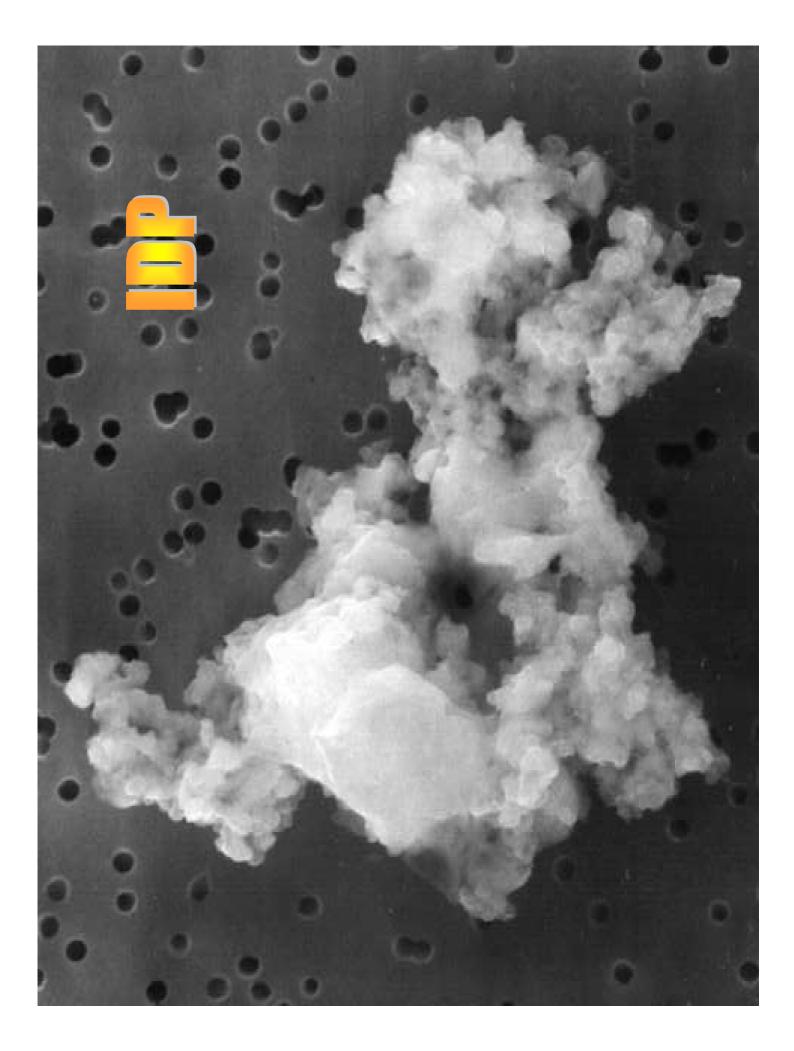


### Murchison



# **Comet Hale-Bopp and M31**





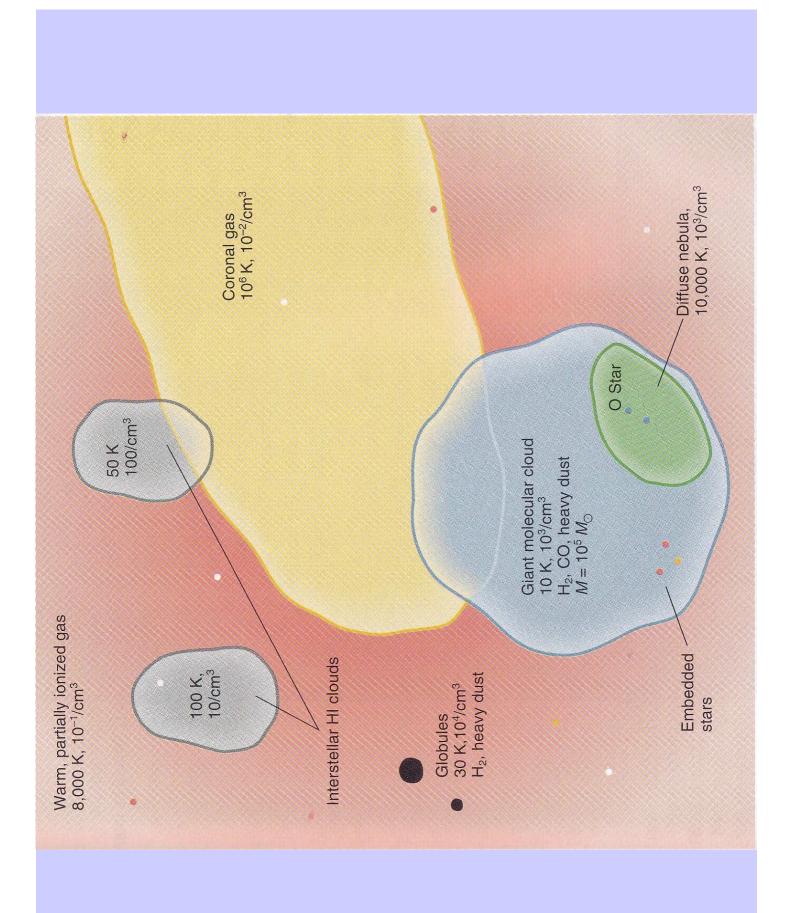


### EARLY EARTH

- Heavy bombardment by comets over 700 mill. years
- Strong geological activity
- First evidence for Life: ~ 3.6 billion years ago



## Fate of the Sun: Planetary Nebula



# **Molecules in the Galactic ISM**

### • Only in cold dense phases:

- contain the bulk of the interstellar gas mass

- composed of atomic and molecular gas diffuse HI clouds  $n_{\rm H} \sim 30 \text{ cm}^{-3}, \text{ T}=100 \text{ K}$ molecular clouds  $n_{\rm H} > 300 \text{ cm}^{-3}, \text{ T}=20 \text{ K}$ 

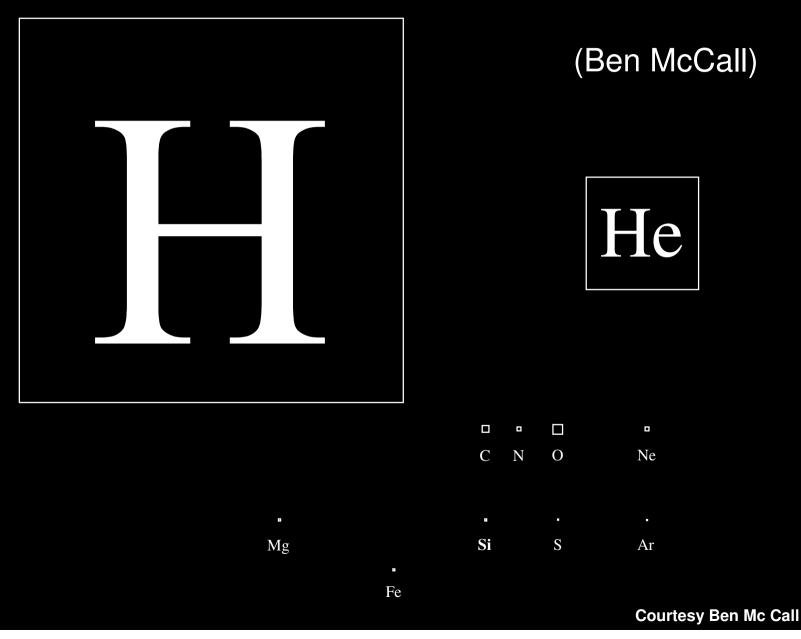
• **Dust grains** 

carbonaceous and silicate particles provide extinction

<u>Astrochemistry of heavy elements</u>

unimolecular and bimolecular gas reactions driven by cosmic rays and UV photons.

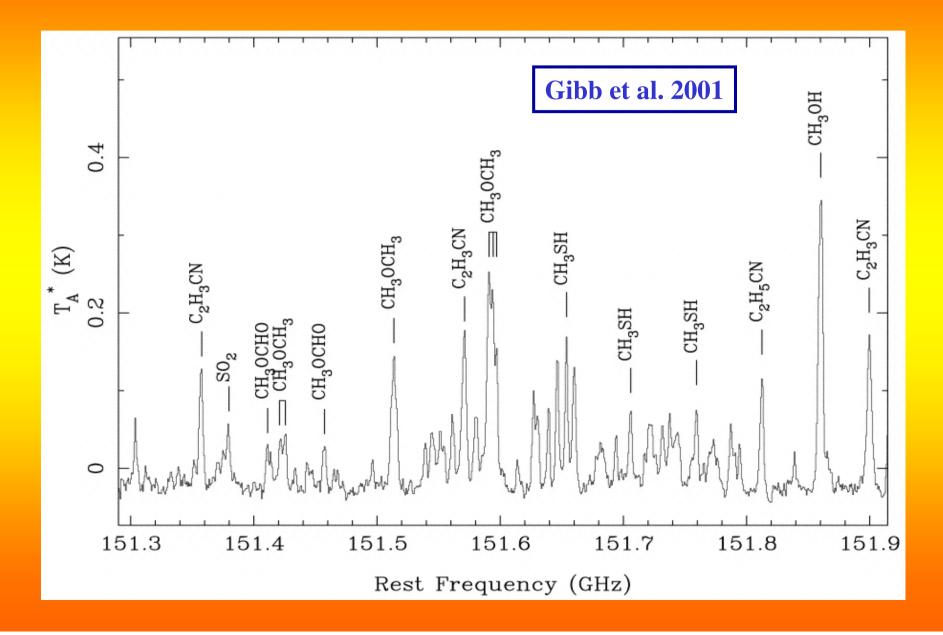
# The Astronomer's Periodic Table



# **Brief History of Astrochemistry**

- 1937-1941: optical detection of CH, CN, CH+
- 1951: Bates & Spitzer theory
- 1968-73: ammonia, formaldehyde, water, CO
- 1970-1980's mm-wave telescopes led to discovery of many species
- Herbst & Klemperer (1973) : ion-molecule chemistry
- 1987: Blake et al. line survey of Orion-KL

### **Survey of organic molecules in G327**



#### 0.20 (1777) 0.20 C<sub>2</sub>H<sub>5</sub>CN 707-616 0.15 -0.15 Glycol-0.10 0.10 TR\* (K) TR\* (K) CH<sub>2</sub>OHCHO CH<sub>2</sub>OHCHO aldehyde 0.05 0.05 0.00 0.00 C<sub>2</sub>H<sub>6</sub>CN 1171533 -0.05 -0.05 71520 71580 71500 71540 71560 71600 93020 93040 93060 93080 Rest Frequency (MHz) Rest Frequency (MHz) the simplest 0.20 0.06 CH<sub>3</sub>OH 817 . 726 Sugar 0.15 N-50 0.04 CH<sub>2</sub>OHCHO TR\* (K) 0.10 0.02 Ta. (N) CH<sub>2</sub>OHCHO 0.05 0.00 U103418 SgrB2(N) CH<sub>2</sub>CHCHO -0.02 0.00 -0.04 -0.05 103360 103380 103400 75320 75340 75380 75400 103420 75360 detected in 2000 Rest Frequency (MHz) Rest Frequency (MHz) 0.4 0.10 toward $C_2S$ CH3OCH3 808 - 717 0.08 0.3 SgrB2(N) 0.05 0.2 TR\* (K) TR- (K) CH<sub>2</sub>OHCHO 0.04 CH<sub>2</sub>OHCHO U82518 0.1 0.02 0.0 0.00

1111

82520

82500

-0.02

103620 103630

103640

103650

Rest Frequency (MHz)

103660

-0.1

. . . .

82440

.........

82480

Rest Frequency (MHz)

82460

909-818

HCOOH

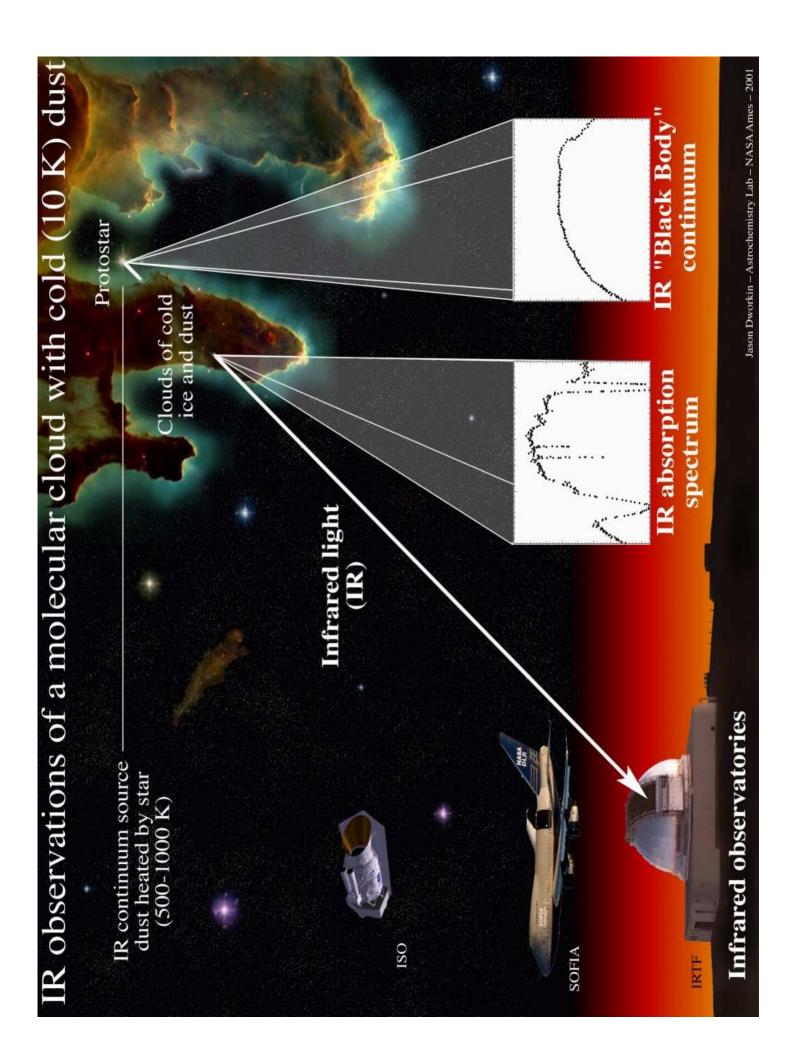
93100

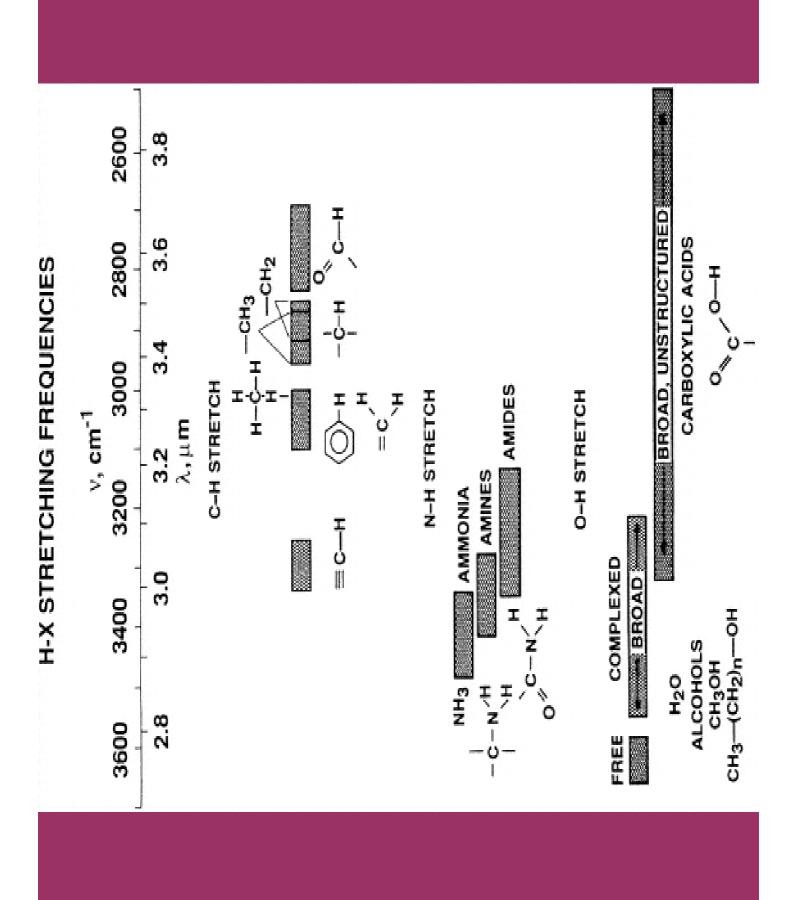
103440

1019 - 928

103670

100.10 - 919





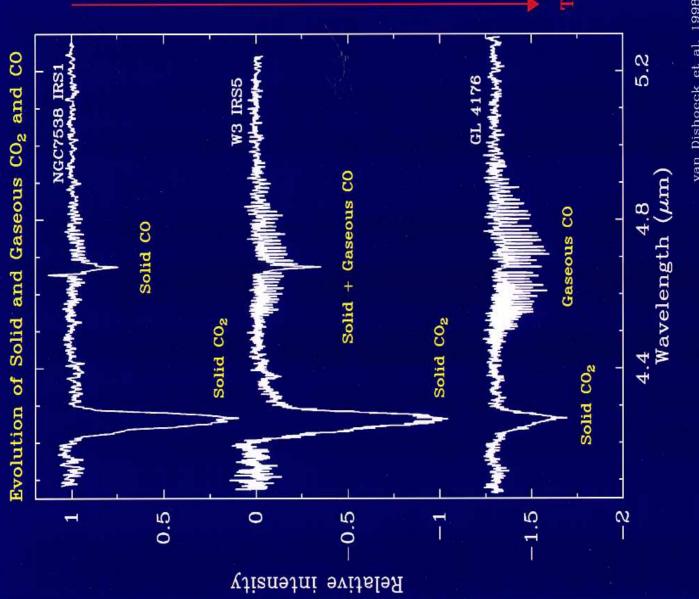
### **THE INFRARED SPACE OBSERVATORY**



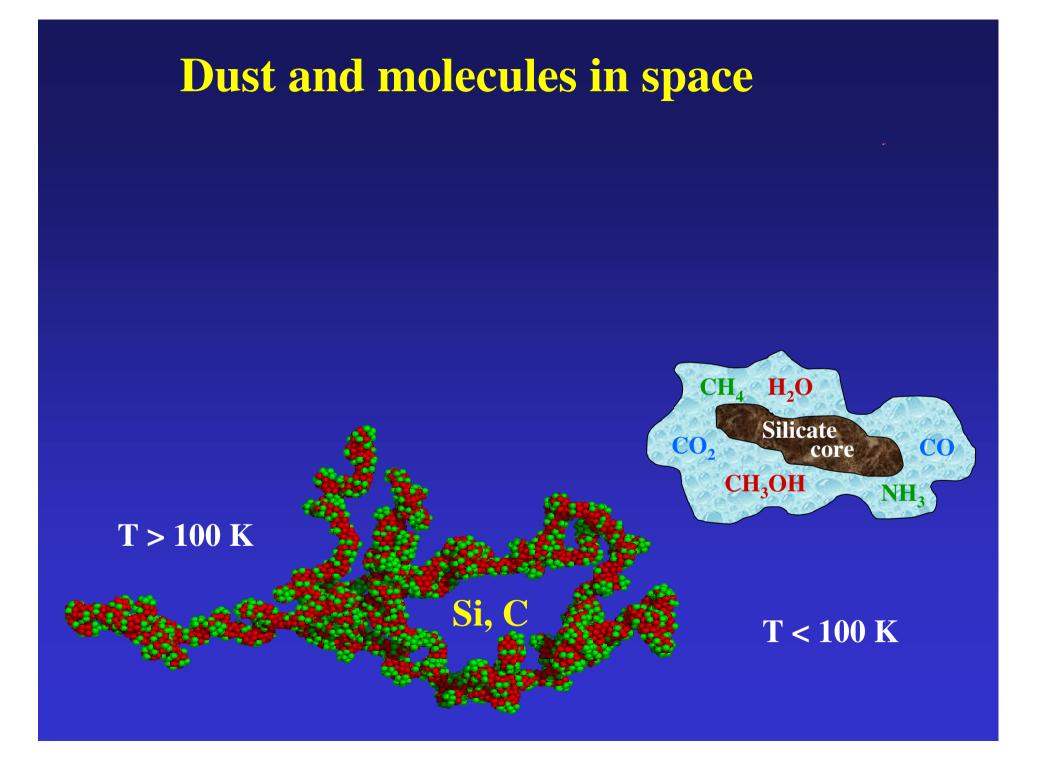
ISO 1995-1998 ....

ISO allowed the first complete inventory of cosmic dust between 2.5-200 micron

- Discovery of new molecules
- Accurate abundances
- Gas-grain interactions
- Irradiation and temperature conditions in proto-stellar environments



van Dishoeck et al. 1998



### **Molecular inventory**

2	3	4	5	6
H <sub>2</sub>	C <sub>3</sub>	c-C₃H	C <sub>5</sub>	C <sub>5</sub> H
AIF	C <sub>2</sub> H	I-C <sub>3</sub> H	C <sub>4</sub> H	I-H <sub>2</sub> C <sub>4</sub>
AICI	C <sub>2</sub> 0	C <sub>3</sub> N	C <sub>4</sub> Si	C <sub>2</sub> H <sub>4</sub>
C <sub>2</sub>	C <sub>2</sub> S	C <sub>3</sub> O	I-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> CN
CH	CH <sub>2</sub>	C <sub>3</sub> S	c-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> NC
CH <sup>+</sup>	HCN	$C_2H_2$	CH <sub>2</sub> CN	CH <sub>3</sub> OH
CN	HCO	$CH_2D^+?$	CH <sub>4</sub>	CH <sub>3</sub> SH
CO	HCO <sup>+</sup>	HCCN	HC <sub>3</sub> N	HC <sub>3</sub> NH <sup>+</sup>
CO <sup>+</sup>	HCS <sup>+</sup>	HCNH <sup>+</sup>	HC <sub>2</sub> NC	HC <sub>2</sub> CHO
CP	HOC <sup>+</sup>	HNCO	нсоон	NH <sub>2</sub> CHO
CSi	H <sub>2</sub> O	HNCS	H <sub>2</sub> CHN	C <sub>5</sub> N
HCI	H <sub>2</sub> S	HOCO <sup>+</sup>	$H_2C_2O$	
KCI	HNC	H <sub>2</sub> CO	H <sub>2</sub> NCN	
NH	HNO	H <sub>2</sub> CN	HNC <sub>3</sub>	
NO	MgCN	H <sub>2</sub> CS	SiH <sub>4</sub>	
NS	MgNC	H <sub>3</sub> O <sup>+</sup>	H <sub>2</sub> COH <sup>+</sup>	
NaCl	$N_2H^+$	NH <sub>3</sub>		
OH	N <sub>2</sub> O	SiC <sub>3</sub>		
PN	NaCN	CH <sub>3</sub>		
SO	OCS			
S0 <sup>+</sup>	SO <sub>2</sub>			
SiN	c-SiC <sub>2</sub>			
SiO	CO <sub>2</sub>			
SiS	$NH_2$			
CS HE	H <sub>3</sub> <sup>+</sup>			
HF	SICN			
FeO?				
SiH	$H_20^+$			
ып				

	8	9	1
	CH <sub>3</sub> C <sub>3</sub> N	CH <sub>3</sub> C <sub>4</sub> H	Cł
N	HCOOCH <sub>3</sub>	CH <sub>3</sub> CH <sub>2</sub> CN	(C
	CH3COOH	(CH <sub>3</sub> ) <sub>2</sub> O	H
	C <sub>7</sub> H	CH <sub>3</sub> CH <sub>2</sub> OH	N
3 3	H <sub>2</sub> C <sub>6</sub>	HC <sub>7</sub> N	
	CH <sub>2</sub> OHCHO	C <sub>8</sub> H	
	C <sub>2</sub> H <sub>6</sub>		

7

C<sub>6</sub>H

CH<sub>2</sub>CHC

CH<sub>3</sub>C<sub>2</sub>H

 $HC_5N$  $HCOCH_3$  $NH_2CH_3$  $c-C_2H_4O$  $CH_2CHOH$ 

#### 

### Physics World, Charnley et al. 2003

Astronomers have made a list of 131 molecules that have been discovered in interstellar space, which range from simple two-atom species (left) to complex molecules that contain up to 13 atoms. Many of these play important roles in terrestrial biochemistry, and several organic classes are represented: acids, aldehydes, ketones, alcohols, ethers, esters and pre-sugars. Some of these molecules, which include structural isomers such as HCN and HNC, are also present in meteorites and in comets. Many of the hydrocarbons that contain multiple carbon atoms exist as long carbon chains. The smallest member of the cyanopolyyne series cyanoacetylene (HC<sub>3</sub>N) - is ubiquitous in molecular clouds, and another member – cyanodecapentayne  $(HC_{11}N)$  – is the largest molecule that has been unambiguously identified in the interstellar medium. A few small ring molecules are present in the list but many larger organic compounds await detection in space. The present authors, for example, are currently using the Arizona Radio Observatory 12 m and Green Bank telescopes to search for ring compounds (PAHs) containing nitrogen. Table courtesy of Al Wootten and updated from www.astrochemistry.net.

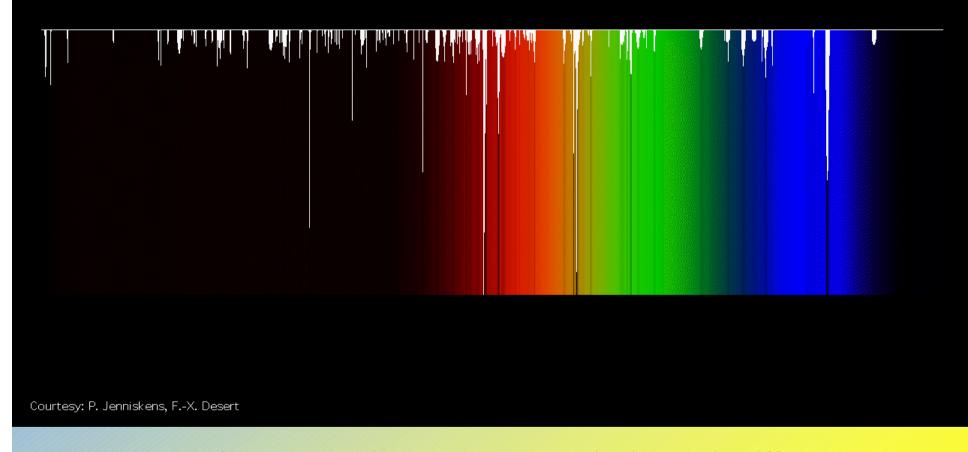
# **Clouds: Compositional Differences**

- Diffuse clouds: atomic ions (C<sup>+</sup>, S<sup>+</sup>), simple molecules (OH, CH, CN), some unexpected polyatomics (C<sub>3</sub>H<sub>2</sub>, C<sub>3</sub>H), plus large unidentifed structures
- Dark Clouds: H<sub>3</sub><sup>+</sup>, CO, long unsaturated carbon chains (e.g. HC<sub>n</sub>N), high D/H
- Cold Prestellar Cores: very high D/H, evidence for selective depletion onto dust
- Hot Protostellar Cores: very high D/H, many complex organics (ethanol, dimethyl ether), evidence for grain chemistry

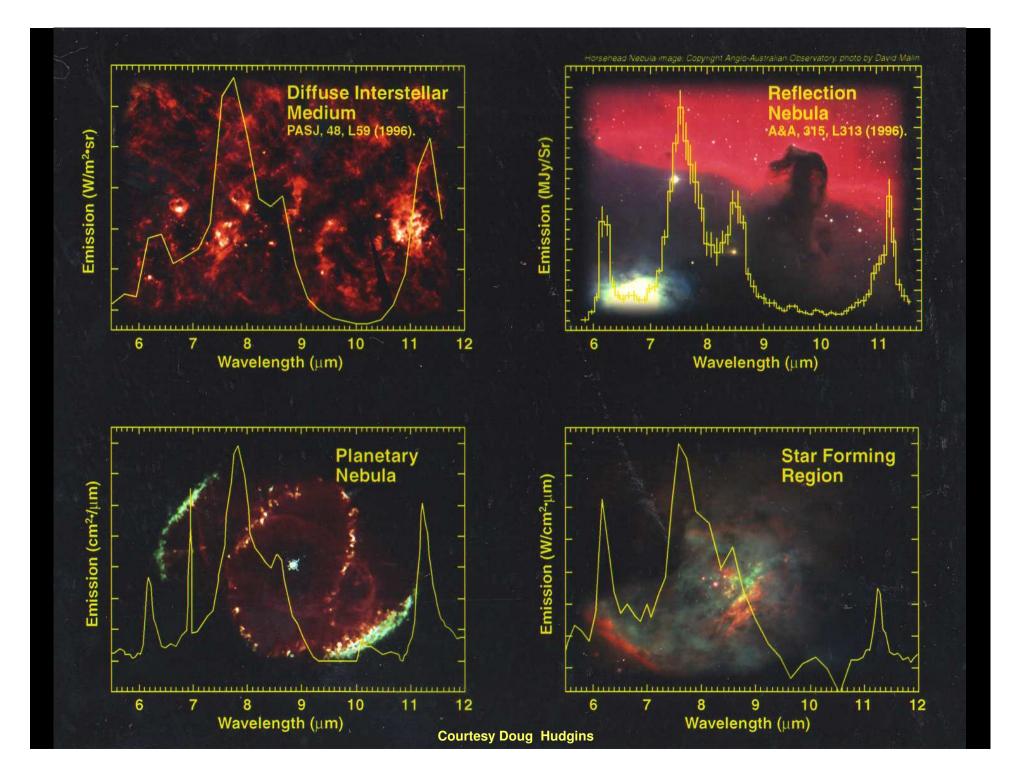
# Astrochemistry or MolecularAstrophysics?

- Chemical diagnostics: gas density (HCO<sup>+</sup>, CS, NH<sub>3</sub>, HCN), molecular hydrogen mass (CO isotopes), temperature (NH<sub>3</sub>, CH<sub>3</sub>OH), infall (H<sub>2</sub>CO), ionization (HCO<sup>+</sup>, DCO<sup>+</sup>), magnetic field (CCS).
- Chemical controls: temperature (CO, H<sub>2</sub>, CII, CI, H<sub>2</sub>O cooling), ambipolar diffusion rate (electron fraction)
- The known molecules have great utility !

### The Diffuse Interstellar Bands



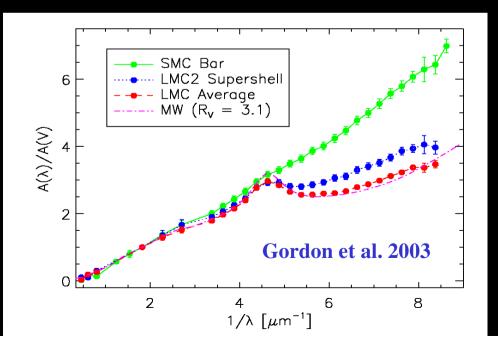
DIBs - Absorption bands which are observed ubiquitously in diffuse clouds Astronomical observations revealed up to ~ 300 DIBs The carrier seems to be \_\_\_\_\_\_ chemically stable \_\_\_\_\_\_ in the gas phase \_\_\_\_\_\_ carbonaceous in nature

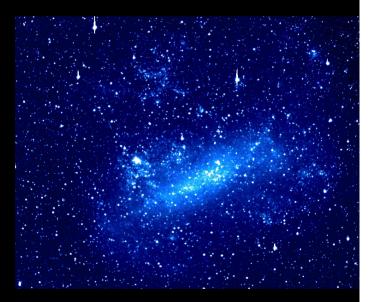


### CARBON CHEMISTRY IN SPACE

The carbon chemistry seems to follow common pathways throughout the Universe

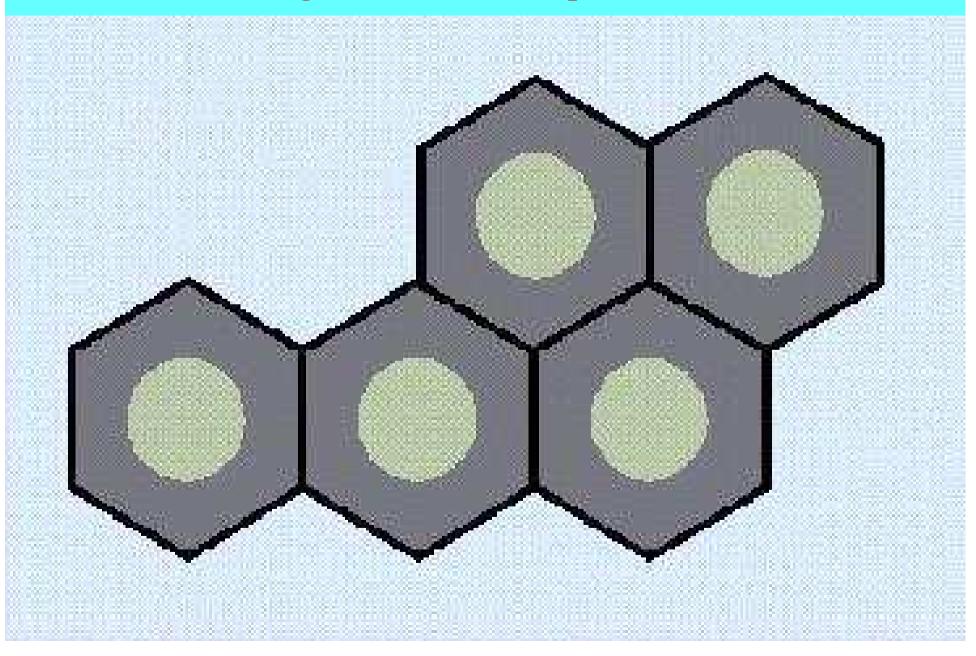
✓ Aromatic hydrocarbons (mid-IR)
✓ Diffuse interstellar bands (Optical)
✓ Aliphatic hydrocarbons (near-IR)
✓ Ices in star-forming regions (near and mid-IR)

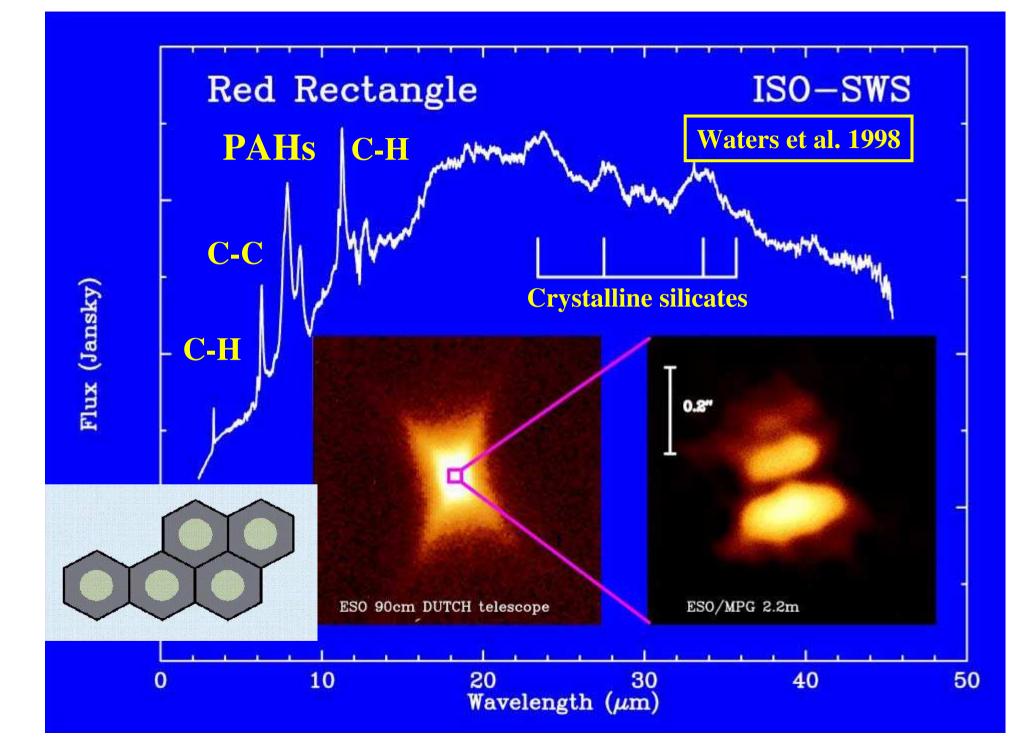




### Interstellar extinction curve → UV bump ~ 220 nm

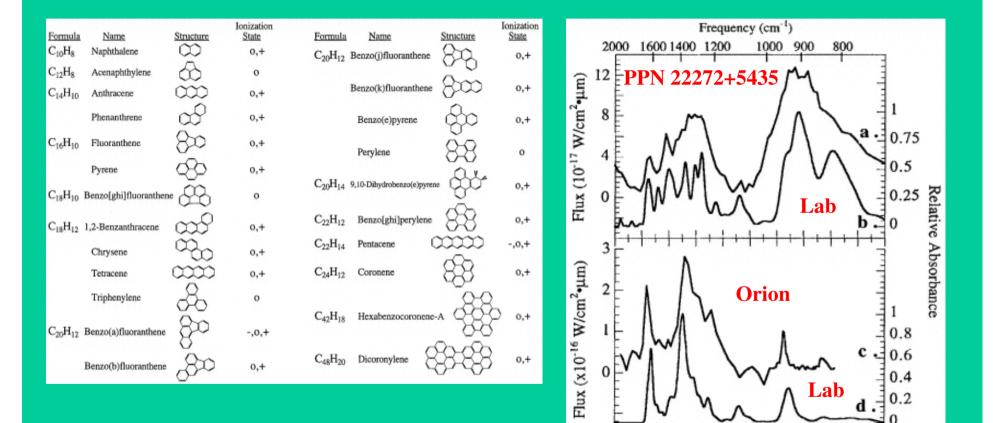
# **Polycyclic Aromatic Hydrocarbons (PAHs): the most abundant organic molecules in space**





### **Polycyclic aromatic hydrocarbons - PAHs**

### **Combined laboratory spectra of neutral and positively charged PAHs**



**IR Database - NASA AMES** 

Allamandola et al. 1999

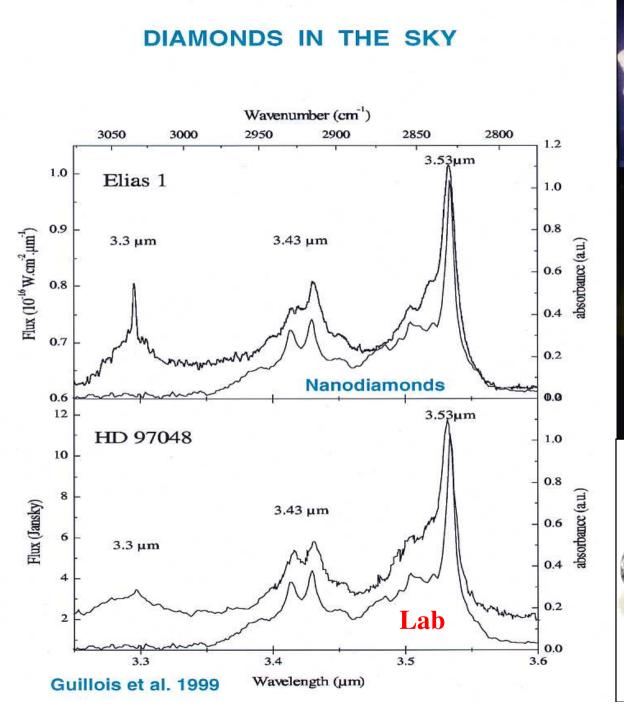
Wavelength (µm)

10 11 12 13 14 15

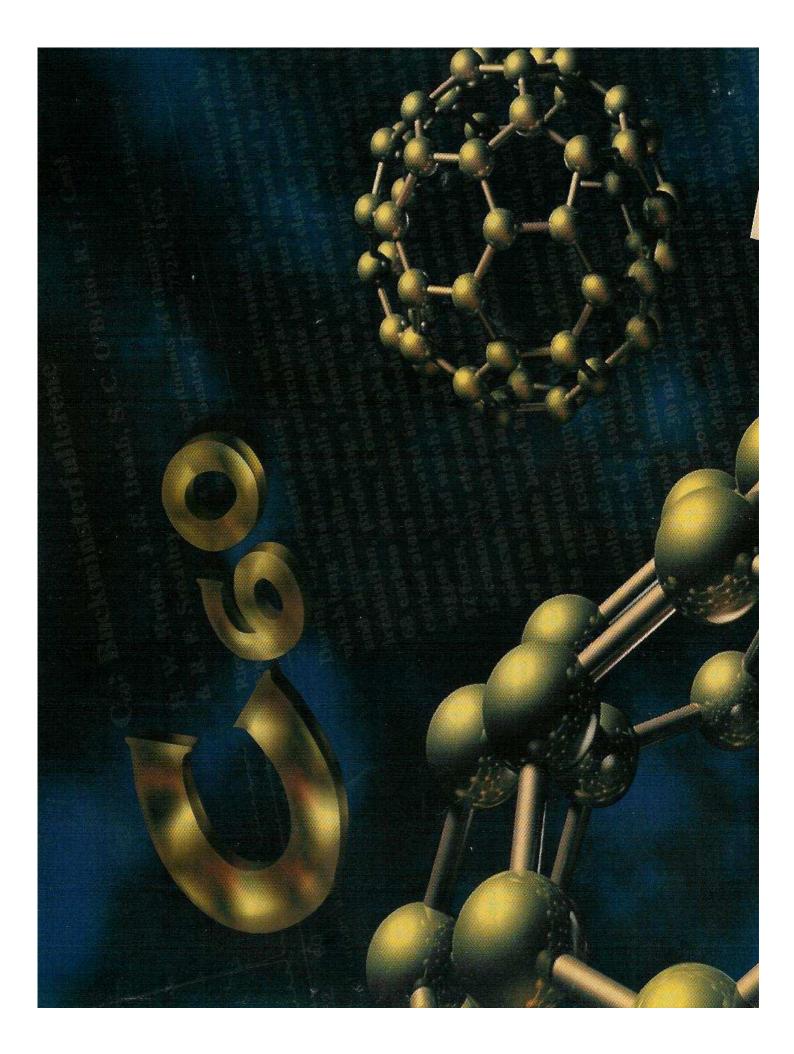
9

7 8

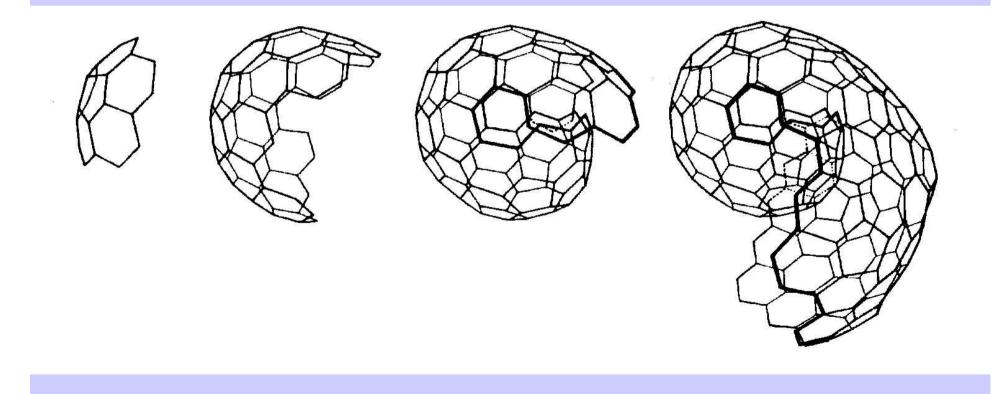
5 6



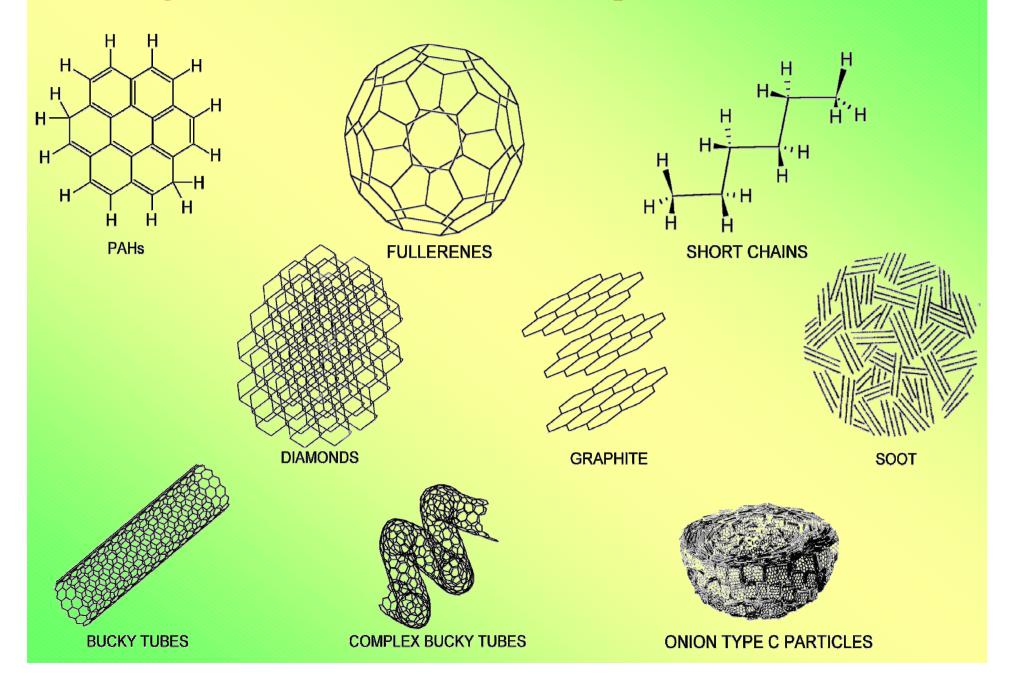




# Nucleation scheme for the formation of fullerene carbon particles in the gas phase



### Large carbonaceous molecules in space (EC2000)



# Summary

- Over 120 molecules detected; most are organic
- Many more await detection
- Laboratory data is key

**Tomorrow:** 

- Possible to trace interstellar material as far as comets

